

Section 1. Product and Company Identification

Product Identification: Mixed Food Diet
 MSDS Number: CRM-MFD
 Company Identification: High-Purity Standards
 P.O. Box 41727
 Charleston, SC 29423
 Telephone: (843) 767-7900
 FAX: (843) 767-7906

In case of emergency call INFOTRAC: 800-535-5053

Section 2. Chemical Composition

Component	CAS/ EINECS Registry #	Percent Concentration	ACGIH TLV	OSHA PEL
Aluminum	7429-90-5/ 231-072-3	<0.001	10 mg/m ³	15 mg/m ³
Arsenic	7440-38-2/ 231-148-6	<0.001	0.01 mg/m ³	10 µg/m ³
Calcium Carbonate	471-34-1/ 207-439-9	0.004 (as Ca)	0.5 mg/m ³	0.5 mg/m ³
Cadmium	7440-43-9/ 231-152-8	<0.001	0.002 mg/m ³ (respirable particulate)	0.005 mg/m ³
Chromium	7440-47-3/ 231-157-5	<0.001	0.5 mg/m ³	1 mg/m ³
Cobalt	7440-48-4/ 231-158-0	<0.001	0.02 mg/m ³	0.1 mg/m ³
Copper	7440-50-8/ 231-159-6	<0.001	0.2 mg/m ³ (fumes)	0.1 mg/m ³ (fumes)
Iron	7439-89-6/ 231-096-4	<0.001	10 mg/m ³	5 mg/m ³
Magnesium	7439-95-4/ 231-104-6	0.001	Not Available	Not Available
Manganese	7439-96-5/ 231-105-1	<0.001	0.2 mg/m ³	C 5 mg/m ³
Molybdenum	7439-98-7/ 231-107-2	<0.001	5 mg/m ³	5 mg/m ³
Nickel	7440-02-0/ 231-111-4	<0.001	1.5 mg/m ³	1 mg/m ³
Ammonium Dihydrogen Phosphate (NH ₄ H ₂ PO ₄)	7722-76-1/ 231-764-5	0.006 (as P)	Not Available	Not Available
Potassium Nitrate (KNO ₃)	7757-79-1/ 231-818-8	0.016 (as K)	Not Available	Not Available
Selenium	7782-49-2/ 231-957-4	<0.001	0.2 mg/m ³	0.2 mg/m ³
Sodium Carbonate (Na ₂ CO ₃)	497-19-8/ 207-838-8	0.006 (as Na)	Not Available	Not Available

Material Safety Data Sheet Mixed Food Diet	Date: January 6, 2006	
CRM-MFD	Rev.: New	Page 2 of 4

Zinc	7440-66-6/231-175-3	<0.001	5 mg/m ³	1 mg/m ³
Nitric Acid	7697-37-2/231-714-2	2	2 mg/kg	5 mg/m ³
Water, deionized	7732-18-5/231-791-2	Balance	Not Available	Not Available

Section 3. Hazard Identification

Emergency Overview: Mildly corrosive. May cause irritation to areas of contact. Wash areas of contact with water for at least 15 minutes. If ingested, do not induce vomiting. Dilute with water and call a physician. While hazardous materials are present at very low concentrations, they may cause cancer.

Target Organs: Eyes, skin, respiratory system, immune system, nasal cavities, teeth, blood, bones. Increases risk of lung, liver, kidney, and bladder cancer with prolonged exposure.

Skin/eye Contact: Liquid may cause burns to skin and eyes.

Inhalation: May cause irritation.

Ingestion: May cause nausea, vomiting, and diarrhea. Ingestion of arsenic compounds may be poisonous, leading to death. Animal studies indicate that prolonged ingestion of some soluble nickel compounds may affect the blood, bone marrow, thymus, spleen, kidneys, and immune system.

Section 4. First Aid Measures

Inhalation: Remove to fresh air. Give artificial respiration if necessary. If breathing is difficult, give oxygen.

Skin/eye Contact: Flush eyes with plenty of water for at least 15 minutes. Remove contaminated shoes and clothing. Rinse affected area with large amount of water followed by washing the area with soap and water. Call a physician if irritation develops.

Ingestion: CALL A PHYSICIAN; If swallowed rinse mouth, do NOT induce vomiting, if conscious give large quantities of water or milk.

Section 5. Fire Fighting Measures

Fire & Explosion hazards: While nitric acid is not combustible, it is a strong oxidizing agent that can react with combustible materials. NO_x compounds can be released in event of fire.

Extinguishing Media: Use any extinguishing media that is suitable for the surrounding area. Use a water spray to dilute nitric acid and to absorb liberated nitrogen oxides.

Specific Methods: Firefighters should wear proper protective equipment and self-contained breathing apparatus with full face piece operated in positive pressure mode.

Section 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Cover the spill with sodium bicarbonate or a soda ash-slaked lime mixture (50:50) to neutralize the acid. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction. Always dispose of in accordance with local regulations.

Material Safety Data Sheet Mixed Food Diet	Date: January 6, 2006	
CRM-MFD	Rev.: New	Page 3 of 4

Section 7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Keep out of direct sunlight and away from heat, water, and incompatible materials. When diluting, the acid should always be added slowly to water and in small amounts. Refer to Section 8 for personal handling instructions. Wash exposed skin area thoroughly after handling.

Section 8. Exposure Controls and Personal Protection

Engineering Controls: No specific controls are needed. Normal room ventilation is adequate.
 Respiratory Protection: Normal room ventilation is adequate.
 Personal Protection: Wear proper gloves, safety glasses with side shields, lab coat/apron.

Section 9. Physical and Chemical Properties

Molecular Weight: N/A
 Boiling Point: Approximately 100°C
 Freezing Point: N/A
 Vapor Pressure (mm): N/A
 Vapor Density (air+1): N/A
 Specific Gravity (H₂O = 1): Approximately 1.0
 Solubility in H₂O: Complete
 Appearance: Colorless to light gray liquid
 Odor: Odorless to a faint pungent odor
 pH: <1

Section 10. Stability and Reactivity

Stability Indicator: YES
 Conditions to Avoid: Metals, chlorine, organic materials, strong alkali, cyanides.
 Incompatibles: Strong reducing agents.
 Hazardous Decomposition Products: NO_x compounds including nitric oxide (NO), nitrogen dioxide (NO₂), nitrous oxide (N₂O) and nitric acid mist or vapor.
 Hazardous Polymerization: NO

Section 11. Toxicological Information

Toxicity Data:
 May affect skin, mucous membranes and eyes. Swallowing may lead to a negative effect on mouth and throat and to the risk of perforation or the corrosion of esophagus and stomach.

RTECS#

HNO ₃ - QU5775000	Al - BD0330000	As- CG0525000	CaCO ₃ - EV9580000
Cd- EU9800000	Co - GF8750000	Cr- GB4200000	Cu- GL5325000
KNO ₃ - TT3700000	Mo - QA4680000	Mn - OO9275000	Na ₂ CO ₃ - VZ4050000
Ni- QR5950000	Se- VS7700000	Zn- ZG8600000	

LD_{LO} Oral, Human: (HNO₃) 430 mg/kg; LD₅₀ Oral, Rat: (Al) >5000 mg/kg; LD₅₀, Oral, Rat: (As) 763 mg/kg; LD_{LO} Oral, Human: (Cd) 2330 mg/kg; LD_{LO} Oral, Rabbit: (Co) 750 mg/kg; LD₅₀ Unreported Route, Rat: (Cr) 27.5 mg/kg; TD_{LO} Oral, Human: (Cu) 120 µg/kg; LD₅₀ Oral, Rat:

Material Safety Data Sheet Mixed Food Diet	Date: January 6, 2006	
CRM-MFD	Rev.: New	Page 4 of 4

(KNO₃) 3750 mg/kg; LD₁₀ Intratracheal, Rabbit: (Mo) 70 mg/kg; LD₅₀ Oral, Rat: (Mn) 9 g/kg; LD₅₀, Oral, Mouse: (Na₂CO₃) 6600 mg/kg; LD₅₀, Intravenous, Mouse: (Ni) 50 mg/kg; LD₅₀, Oral, Rat: (Se) 6700 mg/kg; TD₁₀ Implant; LD₁₀ Oral, Duck: (Zn) 388 mg/kg.

Section 12. Ecological Information

Ecotoxicological information: Do not allow material to reach ground water, water bodies, or sewage system.

Section 13. Disposal Considerations

Follow federal, state and local regulations for waste.

Section 14. Transport Information

D.O.T. Classification: Not hazardous by DOT regulations (based on low concentration of acid).

Section 15. Regulations (Not meant to be all inclusive-selected regulation listed)

TSCA Status: Components of this solution are listed on the TSCA Inventory.

RCRA Status: No

SARA: Subject to the reporting requirements of Section 313 or SARA Title III and of 40 CFR 372

Risk Phrases: R20/21/22, R45 Harmful by inhalation, skin contact, or ingestion. May cause cancer.

Safety Phrases: S36/37/39 Wear suitable protective clothing, gloves and eye/face protection

WHMIS Information (Canada): E: Corrosive

Section 16. Other Information

HPS products are intended for laboratory use only. All products should be handled and used by trained professional personnel only. The responsibility for the safe handling and use of these products rest solely with the buyer and/or user. The MSDS was prepared carefully and represents the best data currently available to us; however, HPS does not certify the data on the MSDS. Certified values for this material are given only on the Certificate of Analysis.

Theodore C. Rains, Ph.D.