

Section 1. Product and Company Identification

Product Identification: ANALCS-R
 MSDS Number: ANALCS-R
 Company Identification: High-Purity Standards
 P.O. Box 41727
 Charleston, SC 29423
 Telephone: (843) 767-7900
 FAX: (843) 767-7906

In case of emergency call INFOTRAC: 800-535-5053

Section 2. Hazard Identification

Classification:

Skin Corrosion/Irritation, Category 1B

Serious Eye Damage/ Eye Irritation, Category 1

Labeling:**Symbol:**

Signal Word: Danger.

Hazard Statement: Causes severe skin burns and eye damage. Causes serious eye damage.

Precautionary Statement: Wear protective gloves/clothing and eye/face protection. Wash thoroughly after handling.

Section 3. Hazard Identification

Component	CAS/EINECS Registry #	Percent Concentration
Antimony	7440-36-0/231-146-5	0.006
Arsenic	7440-38-2/231-148-6	0.001
Barium Carbonate (BaCO ₃)	513-77-9/208-167-3	0.005 (as Ba)
Barium Nitrate (Ba(NO ₃) ₂)	10022-31-8/233-020-5	
Beryllium Acetate (Be ₄ O(C ₂ H ₃ O ₂) ₆)	19049-40-2/242-785-4	0.005 (as Be)
Cadmium	7440-43-9/231-152-8	0.01
Chromium	7440-47-3/231-157-5	0.005
Cobalt	7440-48-4/231-158-0	0.005
Copper	7440-50-8/231-159-6	0.005
Lead	7439-92-1/231-100-4	<0.001
Manganese Acetate Tetrahydrate (Mn(CH ₃ CO ₂) ₂)*4H ₂ O	6156-78-1/211-334-3	0.1 (as Mn)
Nickel	7440-02-0/231-111-4	0.01
Selenium	7782-49-2/231-957-4	<0.001
Silver	7440-22-4/231-131-3	0.002
Thallium	7440-28-0/231-138-1	0.001
Ammonium Metavanadate (NH ₄ VO ₃)	7803-55-6/232-261-3	0.005 (as V)
Zinc	7440-66-6/231-175-3	0.01
Nitric Acid	7697-37-2/231-714-2	2
Hydrofluoric Acid	7664-39-3/231-634-8	<0.001
Water, deionized	7732-18-5/231-791-2	Balance

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**Note: Barium is derived from either Barium carbonate or Barium Nitrate. For this reason both sources are listed on the SDS. Refer to the product's Certificate of Analysis to determine which source was used in the production of the lot number received.*

Section 4. First Aid Measures

IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Gently wash with plenty of soap and water. Rub calcium gluconate gel immediately to skin.

Obtain medical assistance. Wash contaminated clothing before reuse.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTER doctor/physician.

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

IF INHALED: Remove to fresh air and keep at rest in a position comfortable for breathing.

Target Organs: Eyes, skin, respiratory system, teeth, and skeletal system.

Section 5. Fire Fighting Measures

Fire & Explosion hazards: While nitric acid is not combustible, it is a strong oxidizing agent that can react with combustible materials; however, it is present in limited quantities in this solution. NO_x compounds can be released in case of fire. Hydrofluoric acid may ignite or explode on contact with combustible materials.

Extinguishing Media: Use any extinguishing media that is suitable for the surrounding area. Use a water spray to dilute nitric acid and to absorb liberated nitrogen oxides.

Specific Methods: Firefighters should wear proper protective equipment and self-contained breathing apparatus with full face piece operated in positive pressure mode.

Section 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Cover the spill with sodium bicarbonate or a soda ash-slaked lime mixture (50:50) to neutralize the acid. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction. Always dispose of in accordance with local regulations.

Section 7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Keep out of direct sunlight and away from heat, water, and incompatible materials. When diluting, the acid should always be added slowly to water and in small amounts. Refer to Section 8 for personal handling instructions. Wash exposed skin area thoroughly after handling.

Section 8. Exposure Controls and Personal Protection

Engineering Controls: Provide exhaust ventilation or other engineering controls to keep any buildup of airborne contaminants below their respective threshold limit value. Ensure the availability of eyewash stations and safety showers.

Personal Protection: Wear proper gloves, safety glasses with side shields, lab coat/apron.

Exposure Limits:

Component	ACGIH TLV	OSHA PEL
Antimony	0.5 mg/m ³	0.5 mg/m ³
Arsenic	0.01 mg/m ³	10 µg/m ³

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Barium	0.5 mg/m ³	0.5 mg/m ³
Beryllium Acetate	0.002 mg/m ³	0.002 mg/m ³
Cadmium	0.002 mg/m ³ (respirable particulate)	0.005 mg/m ³
Chromium	0.5 mg/m ³	1 mg/m ³
Cobalt	0.02 mg/m ³	0.1 mg/m ³
Copper	0.2 mg/m ³ (fumes)	0.1 mg/m ³ (fumes)
Lead	0.05 mg/m ³	0.05 mg/m ³
Manganese Acetate Tetrahydrate	0.2 mg/m ³	C 5 mg/m ³
Nickel	1.5 mg/m ³	1 mg/m ³
Selenium	0.2 mg/m ³	0.2 mg/m ³
Silver	0.1 mg/m ³	Not Available
Thallium	0.1 mg/m ³	0.1 mg/m ³
Ammonium Metavanadate	0.05 mg/m ³	Not Available
Zinc	5 mg/m ³	1 mg/m ³
Nitric Acid	2 mg/kg	5 mg/m ³
Hydrofluoric Acid	C: 3 mg/ml	2.5 mg/m ³ STEL: 6 mg/ml
Water, deionized	Not Available	Not Available

Section 9. Physical and Chemical Properties

Physical State: Liquid
 Color: Clear, colorless liquid
 Odor: Odorless to a faint pungent odor
 Odor threshold: None
 pH: <2
 Melting point: N/A
 Freezing Point: N/A
 Boiling Point: Approximately 100°C
 Flash point: N/A
 Evaporation rate: N/A
 Flammability: N/A
 Explosion limits: N/A
 Vapor Pressure (mm): N/A
 Vapor Density (air+1): N/A
 Relative density: (H₂O = 1): Approximately 1.0
 Solubility in H₂O: Complete
 Auto ignition temperature: N/A
 Decomposition temperature: N/A
 Molecular Weight: N/A

Section 10. Stability and Reactivity

Stability Indicator: YES
 Conditions to Avoid: Avoid heat and contact with combustible and other incompatible materials.
 Incompatibles: Strong reducing agents, metallic powders, strong bases, chlorine, calcium compounds, hydroxides, organic materials, strong alkali, cyanides.

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Hazardous Decomposition Products: HF and NO_x compounds including nitric oxide (NO), nitrogen dioxide (NO₂), nitrous oxide (N₂O) and nitric acid mist or vapor.

Hazardous Polymerization: Will not occur.

Section 11. Toxicological Information

May affect skin, mucous membranes and eyes. Swallowing may lead to a negative effect on mouth and throat and to the risk of perforation or the corrosion of esophagus and stomach.

RTECS#

HNO ₃ - QU5775000	HF- MW7875000	
Sb-CC4025000	As- CG0525000	BaCO ₃ - CQ8600000
Ba(NO ₃) ₂ - CQ9625000	Be- DS1750000	Cd- EU9800000
Cr- GB4200000	Co- GF8750000	Cu- GL5325000
Pb- OF7525000	Mn- AI5775000	Ni- QR5950000
Se- VS7700000	Ag- VW3500000	Tl- XG3425000
V- YW0875000	Zn- ZG8600000	

LD_{LO} Oral, Human: (Nitric Acid) 430 mg/kg;

LC_{LO} Inhalation, Human: (Hydrofluoric Acid) 50 mg/kg/30 min;

LD₅₀ Oral, Rat: (Antimony) 7g/kg;

LD₅₀, Oral, Rat: (Arsenic) 763 mg/kg;

LD_{LO} Oral, Human: (Barium Carbonate) 17 mg/kg;

LD₅₀ Oral, Rat: (Ba(NO₃)₂) 355 mg/kg.

TD_{LO} Intratracheal, Rat: (Beryllium Acetate) 13 mg/kg;

LD_{LO} Oral, Human: (Cadmium) 2330 mg/kg;

LD₅₀ Unreported Route, Rat: (Chromium) 27.5 mg/kg;

LD_{LO} Oral, Rabbit: (Cobalt) 750 mg/kg;

TD_{LO} Oral, Human: (Copper) 120 µg/kg;

TD₅₀ Oral, Woman: (Lead) 450 mg/kg/6 year;

LD₅₀ Oral, Rat: (Manganese) 3730mg/kg;

LD₅₀, Intravenous, Mouse: (Nickel) 50 mg/kg;

LD₅₀, Oral, Rat: (Selenium) 6700 mg/kg;

TD_{LO} Implant, Mouse: (Silver) 11 g/kg; LD₅₀,

TD_{LO} Oral, Man: (Thallium) 5,714 µg/kg;

LD₅₀ Oral, Rat: (Ammonium Metavanadate) 58,100 µg/kg;

LD_{LO} Oral, Duck: (Zinc) 388 mg/kg.

Section 12. Ecological Information

Ecotoxicological information: Do not allow material to reach ground water, water bodies, or sewage system. High concentrations of zinc have been shown to be detrimental to aquatic life. Beryllium and its compounds are considered to have high acute and chronic toxicity to aquatic life. Beryllium is more toxic in soft water than in hard water.

Section 13. Disposal Considerations

Follow federal, state and local regulations for acid waste.

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Section 14. Transport Information

D.O.T. Classification: Hazardous by IATA and 49CFR regulations (based on concentration of acid).

D.O.T. Shipping Name: Corrosive liquid, Acidic, Inorganic, n.o.s. (Nitric Acid Solution)

D.O.T. Hazard Class: 8

U.N./N.A. Number: 3264

Packing Group: II

D.O.T. Label: Corrosive (8)

Section 15. Regulations (Not meant to be all inclusive-selected regulation listed)

TSCA Status: Components of this solution are listed on the TSCA Inventory.

RCRA Status: Yes (Hydrofluoric Acid-U134)

SARA: Subject to the reporting requirements of Section 313 of SARA Title III and of 40 CFR 372

Risk Phrases: R20/21/22, R24, R25, R34 Harmful by inhalation, skin contact, or if ingested.

Toxic in contact with skin. Toxic if swallowed. Causes burns.

Safety Phrases: S24, S25, S36/37/39 Avoid contact with the skin. Avoid contact with eyes. Wear suitable protective clothing, gloves and eye/face protection

WHMIS Information (Canada): E: Corrosive

Section 16. Other Information

HPS products are intended for laboratory use only. All products should be handled and used by trained professional personnel only. The responsibility for the safe handling and use of these products rests solely with the buyer and/or user. The SDS was prepared carefully and represents the best data currently available to us; however, HPS does not certify the data on the SDS. Certified values for this material are given only on the Certificate of Analysis.

Theodore C. Rains, Ph.D.