

## Section 1. Product and Company Identification

Product Identification: AAS-SSWS  
 MSDS Number: AAS-SSWS  
 Recommended Use: For Laboratory Use  
 Company Identification: High-Purity Standards  
 P.O. Box 41727  
 Charleston, SC 29423  
 Telephone: (843) 767-7900  
 FAX: (843) 767-7906

In case of emergency call INFOTRAC: 800-535-5053

## Section 2. Hazard Identification

**Classification:**

Skin Corrosion/Irritation, Category 1B  
 Serious Eye Damage/ Eye Irritation, Category 1

**Labeling:****Symbol:**

**Signal Word:** Danger.

**Hazard Statement:** Causes severe skin burns and eye damage. Causes serious eye damage. Suspected of causing cancer.

**Precautionary Statement:** Wear protective gloves/clothing and eye/face protection. Wash thoroughly after handling.

## Section 3. Composition

Component	CAS/EINECS Registry #	Percent Concentration
Antimony	7440-36-0/231-146-5	0.001
Arsenic	7440-38-2/231-148-6	<0.001
Cadmium	7440-43-9/231-152-8	<0.001
Lead	7439-92-1/231-100-4	<0.001
Selenium	7782-49-2/231-957-4	<0.001
Thallium	7440-28-0/231-138-1	<0.001
Nitric Acid (HNO <sub>3</sub> )	7697-37-2/231-714-2	2
Hydrofluoric Acid	7664-39-3/231-634-8	0.05
Water, deionized	7732-18-5/231-791-2	Balance

## Section 4. First Aid Measures

**IF ON SKIN (or hair):** Remove/Take off immediately all contaminated clothing. Gently wash with plenty of soap and water. Rub calcium gluconate gel immediately to skin. Obtain medical assistance. Wash contaminated clothing before reuse.

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**IF IN EYES:** Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Immediately call a POISON CENTER doctor/physician.

**IF SWALLOWED:** Rinse mouth. Do NOT induce vomiting.

**IF INHALED:** Remove to fresh air and keep at rest in a position comfortable for breathing.

**Target Organs:** Eyes, skin, respiratory system, teeth, and skeletal system.

### Section 5. Fire Fighting Measures

**Fire & Explosion hazards:** While nitric acid is not combustible, it is a strong oxidizing agent that can react with combustible materials; however, it is present in limited quantities in this solution. NO<sub>x</sub> compounds can be released in case of fire.

**Extinguishing Media:** Use any extinguishing media that is suitable for the surrounding area. Use a water spray to dilute nitric acid and to absorb liberated nitrogen oxides.

**Specific Methods:** Firefighters should wear proper protective equipment and self-contained breathing apparatus with full face piece operated in positive pressure mode.

### Section 6. Accidental Release Measures

Ventilate area of leak or spill. Wear appropriate personal protective equipment as specified in Section 8. Do not allow to enter drainage systems or water ways. Dike area and dilute spill with water and neutralize with soda ash, limestone, etc. Place the neutralized material into containers suitable for eventual disposal, reclamation, or destruction. Always dispose of in accordance with local regulations.

### Section 7. Handling and Storage

Store in a cool, dry, ventilated storage area with acid resistant floors and good drainage. Keep out of direct sunlight and away from heat, water, and incompatible materials. When diluting, the acid should always be added slowly to water and in small amounts. Refer to Section 8 for personal handling instructions.

### Section 8. Exposure Controls and Personal Protection

**Engineering Controls:** Provide exhaust ventilation or other engineering controls to keep any buildup of airborne contaminants below their respective threshold limit value. Ensure the availability of eyewash stations and safety showers.

**Personal Protection:** Wear proper gloves, safety glasses with side shields, lab coat/apron.

**Exposure Limits:**

<b>Component</b>	<b>ACGIH TLV</b>	<b>OSHA PEL</b>
Antimony	0.5 mg/m <sup>3</sup>	0.5 mg/m <sup>3</sup>
Arsenic	0.01 mg/m <sup>3</sup>	10 µg/ m <sup>3</sup>
Cadmium	0.002 mg/m <sup>3</sup> (respirable particulate)	0.005 mg/m <sup>3</sup>
Lead	0.05 mg/m <sup>3</sup>	0.05 mg/m <sup>3</sup>
Selenium	0.2 mg/m <sup>3</sup>	0.2 mg/ m <sup>3</sup>
Thallium	0.1 mg/m <sup>3</sup>	0.1 mg/m <sup>3</sup>
Nitric Acid	2 mg/kg	5 mg/m <sup>3</sup>
Hydrofluoric Acid	C: 3 mg/ml	2.5 mg/m <sup>3</sup> STEL: 6 mg/ml
Water, deionized	Not Available	Not Available

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### Section 9. Physical and Chemical Properties

Physical State: Liquid  
Color: Colorless to light gray  
Odor: Odorless to a faint pungent odor  
Odor threshold: None  
pH: <2  
Melting point: N/A  
Freezing Point: N/A  
Boiling Point: Approximately 100°C  
Flash point: N/A  
Evaporation rate: N/A  
Flammability: N/A  
Explosion limits: N/A  
Vapor Pressure (mm): N/A  
Vapor Density (air+1): N/A  
Relative density: (H<sub>2</sub>O = 1): Approximately 1.0  
Solubility in H<sub>2</sub>O: Complete  
Auto ignition temperature: N/A  
Decomposition temperature: N/A  
Molecular Weight: N/A

### Section 10. Stability and Reactivity

Stability Indicator: Decomposes slowly to release oxygen.  
Conditions to Avoid: Metals, chlorine, organic materials, strong alkali, cyanides.  
Incompatibles: Strong reducing agents.  
Hazardous Decomposition Products: NO<sub>x</sub> compounds including nitric oxide (NO), nitrogen dioxide (NO<sub>2</sub>), nitrous oxide (N<sub>2</sub>O) and nitric acid mist or vapor.  
Hazardous Polymerization: Will not occur.

### Section 11. Toxicological Information

May affect skin, mucous membranes and eyes. Swallowing may lead to a negative effect on mouth and throat and to the risk of perforation or the corrosion of esophagus and stomach.

Arsenic, Cadmium, and Lead are known or investigated as possible carcinogenic substances

**RTECS#**

HNO <sub>3</sub> : QU5775000	HF: MW7875000	Sb: CC4025000
As: CG0525000	Cd: EU9800000	Pb: OF7525000
Se: VS7700000	Tl: XG3425000	

**Toxicity Data:**

LD<sub>LO</sub> Oral, Human: (Nitric Acid) 430 mg/kg  
LC<sub>LO</sub> Inhalation, Human: (Hydrofluoric Acid) 50 mg/kg/30 min  
LD<sub>50</sub> Oral, Rat: (Antimony) 7g/kg  
LD<sub>50</sub>, Oral, Rat: (Arsenic) 763 mg/kg  
LD<sub>LO</sub> Oral, Human: (Cadmium) 2330 mg/kg  
TD<sub>50</sub> Oral, Woman: (Lead) 450 mg/kg/6 year  
LD<sub>50</sub>, Oral, Rat: (Selenium) 6700 mg/kg

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TD<sub>LO</sub> Oral, Man: (Thallium) 5,714 µg/kg.

#### Section 12. Ecological Information

Ecotoxicological information: Do not allow material to reach ground water, water bodies, or sewage system.

#### Section 13. Disposal Considerations

General: Follow Federal, state and local regulations for waste.

#### Section 14. Transport Information

D.O.T. Classification: Hazardous by IATA and 49CFR regulations (based on concentration of acid).

D.O.T. Shipping Name: Corrosive liquid, Acidic, Inorganic, n.o.s. (Nitric Acid Solution)

D.O.T. Hazard Class: 8

U.N./N.A. Number: 3264

Packing Group: II

D.O.T. Label: Corrosive (8)

#### Section 15. Regulations (Not meant to be all inclusive-selected regulation listed)

TSCA Status: Components of this solution are listed on the TSCA Inventory.

RCRA Status: Yes (Hydrofluoric Acid-U134)

SARA: Subject to the reporting requirements of Section 313 or SARA Title III and of 40 CFR 372

Risk Phrases: R20/21/22. Harmful by inhalation, skin contact, or if swallowed.

Safety Phrases: S36/37/39. Wear suitable protective clothing, gloves and eye/face protection.

WHMIS Information (Canada): E: Corrosive

#### Section 16. Other Information

HPS products are intended for laboratory use only. All products should be handled and used by trained professional personnel only. The responsibility for the safe handling and use of these products rests solely with the buyer and/or user. The SDS was prepared carefully and represents the best data currently available to us; however, HPS does not certify the data on the SDS. Certified values for this material are given only on the Certificate of Analysis.

Theodore C. Rains, Ph.D.